

## Silicon versus germanium – A historical perspective

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- The first transistor was made of germanium (Ge).
- However, discrete transistors soon transitioned from Ge to Si.

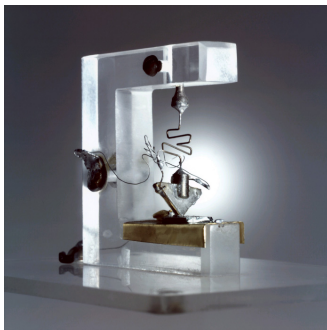
### Advantages of Ge:

- Crystal growth possible at lower temperatures.
- Substantially higher electron and hole mobility indicating potential for higher-speed devices.

### Advantages of Si:

- Stable and strong material with same crystal structure as diamond.
- Excellent oxide ( $\text{SiO}_2$ ). Thermal oxide forms an excellent interface with Si; the interface has a very low interface-state density ( $\sim 10^{10} \text{ cm}^{-2} \text{ eV}^{-1}$ ). Note that *deposited*  $\text{SiO}_2$  does *not* produce the excellent interface that the thermal oxide forms. Also note that this advantage has less significance for FETs using high- $\epsilon_r$  (high k) gate dielectric materials (e.g.  $\text{HfO}_2$ ).
- Si has a larger bandgap and thus becomes intrinsic at higher temperatures.
- Si is less expensive due to the greater abundance of element. The major raw material for Si wafer fabrication is *sand* and there is lots of sand ( $\text{SiO}_2$ ) available.

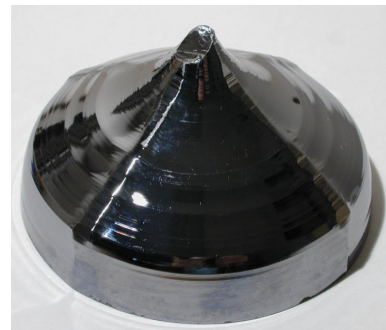
The following story illustrates as to why the transition from Ge to Si was necessary: The first transistor and the first commercial transistors were bipolar junction transistors and they were made of Ge. Motorola Corporation was a manufacturer of car radios. Motorola was one of the first companies to make transistor radios. Naturally the transistors were made of Ge. However, if a car was parked in the sun on a really hot summer day, the radio no longer worked. Why? Because Ge had become became intrinsic; this caused the n-type and p-type regions to lose their distinct properties and, as a result, the bipolar junction transistors no longer worked. This gave Motorola a strong motivation to replace Ge bipolar transistor with Si bipolar transistors.



First transistor (1948)



Early "Motorola Chrome Nose" radio



Si wafer boule